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ANSWER: ANSWER: 3.6 mol of  $\text{Fe}_2\text{O}_3$  0.9 mol of  $\text{Fe}_2\text{O}_3$  2 mol of  $\text{Fe}_2\text{O}_3$  1.8 mol of  $\text{Fe}_2\text{O}_3$  The amount of is the amount of. four times one-quarter twice one-half equal to Typesetting math: 75% 3/31/2019 Ch 9 HW 3/15 Correct Part B How many moles of hydrogen gas would be needed to react with excess carbon dioxide to produce 88.1 of water vapor?

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